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<p>(54) Title: DETERGENT COMPOSITIONS</p> <p>(57) Abstract</p> <p>A tablet of compacted particulate detergent composition including surfactant and detergency builder, suitable for fabric washing, is sub-divided into a plurality of discrete regions with differing compositions. At least one region of the tablet contains a water-swellable polymer serving to promote disintegration of this region. The polymer is present in this region in a concentration greater than in at least one other region of the tablet so as to cause disintegration of the region at a greater speed than disintegration of another region of the tablet.</p>			

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DETERGENT COMPOSITIONS

This invention relates to cleaning compositions in the form of tablets, eg. for use in fabric washing or machine dishwashing.

5      Detergent compositions in tablet form are described, for example, in US 3953350 (Kao) and EP-A-711827 (Unilever), and are sold commercially in Spain. Tablets for machine dishwashing are described in WO96/28530 (P&G). Tablets have advantages over powdered products in that they do not 10     require measuring and are thus easier to handle and dispense into the washload.

Tablets of a cleaning composition are generally made by compressing or compacting a quantity of the composition in particulate form. It is desirable that tablets have 15     adequate strength when dry, yet disperse and dissolve quickly when added to wash water. In such tablets, any surfactant functions as a binder, plasticising the tablet. However, it can also retard disintegration of the tablet by forming a viscous gel when the tablet comes into contact 20     with water.

There have been a number of proposals for tablets which are subdivided into separate regions (eg. layers) which differ in their composition. This has been done in order to isolate components of the tablet composition during storage

of the tablet.

GB-A-911204 discloses tablets in which a detergent tablet contains peroxygen bleach and a bleach activator is confined to a separate layer of the tablet or insert within

5 it.

GB-A-1423536 discloses tablets in which bleach activator is contained in separate, polymer-coated granules while the main part of the tablets includes a soluble or swellable starch.

10 EP-A-481793 discloses tablets in which sodium percarbonate is isolated in a discrete region of the tablet and enhance stability.

15 US-A-3962107 discloses a denture cleaning tablet in which enzyme is contained in one layer of the tablet while peroxygen bleach is contained in another, slower-dissolving layer. Both layers contain effervescence producing materials.

20 US-A-4099912 teaches that fabric washing should be performed using a plurality of tablets containing a plurality of different detergent composition components.

The present invention makes use of a water-swellable polymeric material to accelerate the disintegration of at

least one region of a tablet, ahead of another region or regions.

According to the present invention there is provided a tablet of compacted particulate cleaning composition

5 containing surfactant and detergency builder, and which has a plurality of discrete regions with differing compositions, characterised in that at least one said region of the tablet contains water-swellable polymer in a concentration greater than in at least one other region of  
10 the tablet, to promote disintegration of the first said region(s) before the said other(s).

The regions of the tablet are most likely to be separate layers within a tablet. However, a discrete region of a tablet could be a core or an insert while another discrete  
15 region could be a shell or coating around such core or insert.

The first said region(s) of the tablets which disintegrate and dissolve ahead of other regions can contain an ingredient (eg. enzyme) which is intended to function in  
20 the wash liquor before another constituent, contained in the other region(s) of the tablets, is fully released into the wash liquor.

Another possibility is that the other region(s) of the tablet which disintegrate more slowly contain an ingredient  
25 which is intended to function towards the end of the wash

cycle. For example this could be an antiredeposition polymer or soil suspending polymer. If this is released later in the wash cycle, the proportion of it which is taken up in surfactant micelles and effectively wasted will

5 be minimised.

A further possibility is that a tablet has two layers dissolving at different rates which contain materials to give different values of pH in the wash liquor at different times during a wash cycle.

10 A tablet of this invention may be intended for use in machine dishwashing. Such a tablet is likely to contain surfactant in a low concentration such as 0.5 to 2 wt% based on the whole tablet, although higher concentrations ranging up to 10 wt% may be used. Such will typically

15 contain salts, such as over 60 wt%, often over 85 wt% of the tablet.

Water soluble salts typically used in machine dishwashing compositions are phosphates (including condensed phosphates) carbonates and silicates, generally as alkali metal salts. Water soluble alkali metal salts selected from phosphates, carbonates and silicates may provide 20 60 wt% or more of a dishwashing composition.

In such a machine dishwashing tablet, a discrete region which disintegrates and dissolves first may be a pre-wash

composition containing enzyme(s) and some water soluble salts.

Another possibility is that a tablet of this invention will be intended for fabric washing. In this event the tablet  
5 ~~will be likely to contain at least 2 wt%, probably at least 5 wt%, up to 40 or 50 wt% surfactant based on the whole tablet, and from 5 to 8 wt% detergency builder, based on the whole tablet.~~

10 In such a tablet a discrete region which disintegrates and dissolves first may be a pre-wash composition containing enzyme(s). Such a pre-wash composition will generally include detergency builder as 5 to 90% by weight of the discrete region.

15 The remainder of the composition may then be suitable for the main fabric wash, containing from 5, preferably 10 wt% up to 40% or 50% surfactant.

20 It will be appreciated that within this invention it is possible that the first said region, containing water-swellable polymeric material, does not itself contain surfactant or detergency builder. It could for example contain enzyme(s) and water-soluble carrier salts which have no function as water-softeners.

However, the first said region(s) which contains water-

swellable polymer will often contain detergency builder salts and may contain at least some surfactant.

For instance, for fabric washing, a pre-wash composition, used to form a discrete region of a tablet, might contain 5 enzyme(s), 5 to 90 wt% detergency builder and from 0 to 2% surfactant, together with at least 5 wt% of particles of water-swellable polymer.

In a second aspect, this invention provides a tablet of compacted particulate detergent composition containing 10 surfactant and detergency builder, which has a plurality of discrete regions with differing compositions, characterised in that at least one region of the tablet has a composition which causes it to disintegrate ahead of at least one other region of the tablet,

15 wherein the first said region contains enzyme in a quantity greater than in the other said region, and the other said region contains surfactant and/or bleach in a greater quantity than in the first said region.

The disintegration of the first said region is preferably 20 promoted by incorporation of a water-swellable polymer. However, it could be promoted by incorporation of water soluble materials, more specifically by particles containing at least 40% (by weight of these particles (ii)) of one or more materials selected from 25 • compounds with a water-solubility exceeding 50

grams per 100 grams water

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- phase I sodium tripolyphosphate or
- sodium tripolyphosphate which is partially hydrated so as to contain water of hydration in an amount which is at least 0.5% by weight of the sodium tripolyphosphate in the particles.

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There may be a substantial difference in the times for disintegration of the first said region(s) and the other said region(s). An addition of a tablet to water at 20°C the said other region(s) may remain intact for at least 5 or even at least 10 minutes after disintegration of the first said region(s).

Materials which may be used in tablets of this invention will now be discussed in more detail.

15 The water-swellable polymer

Suitable water-swellable polymeric materials are usually water-insoluble. Preferably they have sufficient water-absorptivity that they can absorb at least four times their own weight of water, ie. a water uptake of at least 4gm per gm.

A number of such materials are known, and are generally based on cellulose which may be chemically modified to enhance its water uptake capacity. Sometimes such modified celluloses have ionic substituents but for this invention

it is preferred that any substituents are nonionic.

Surprisingly, we have found that such a material is more effective if it has a relatively large particle size. We therefore prefer that the polymeric material has a particle dimension of at least 400 better at least 500 micrometres.

Such polymeric material with a particle dimension of at least 400 micrometres is preferably an agglomerate of smaller particles whose largest dimension is no greater than 200 micrometres, better no greater than 150

10 micrometres. This makes it possible for at least some of the polymer particles to break up during a wash cycle.

The material may exist as relatively rounded particles, or as relatively flat particles such as flakes or discs. In the latter case a dimension (diameter) of the flakes will 15 be larger, perhaps substantially larger, than the diameter of a sphere with the same volume.

The largest dimension of particles of the polymeric material may be determined by sieve analysis, and the shape of the particles can be observed under a microscope.

20 It is customary to use sodium carboxymethylcellulose (SCMC) in detergent compositions, usually as not more than 3 wt% of the composition. We have found that such quantities of SCMC are generally ineffective to promote disintegration.

We have found it desirable to use swellable polymeric materials with little or no ionic character. Such materials may be polysaccharides with little or no ionic substitution.

5 The absence or near absence of ionic substitution can be expressed by stating that the charge density of the polymeric material is low, such as less than  $10^{-3}$ , better less than  $6 \times 10^{-4}$  or even zero. The term "charge density" denotes the number of charges on a polymer molecule divided  
10 by the molecular weight of the polymer. It is essentially the same as the average number of charges on a repeat unit of the polymer divided by the average molecular weight of a repeat unit.

15 The water-insoluble, water-swellable material is preferably added as particles which contain such material as at least 75% of the anhydrous weight of these particles (i.e ignoring their moisture content) and usually they will contain little or nothing except the polymer and any accompanying moisture.

20 These particles preferably do not provide more than 5 or 8 wt% of the tablet as a whole. However, a discrete region of a tablet may contain a higher concentration of such particles, e.g. from 3 wt% up to 10 or 15 wt% of the region.

Further disintegration-promoting particles

The water-swellable polymer may be accompanied in the first said region(s) of the tablet by further particles which contain at least 40% of their own weight, better at least 5 50%, of a material which has a solubility in deionised water at 20°C of at least 50 grams per 100 grams of water.

The said particles may provide material of this specified solubility in an amount which is from 5 to 40 wt% of the first said region(s) of the tablet.

10 A solubility of at least 50 grams per 100 grams of water at 20°C is an exceptionally high solubility: many materials which are classified as water soluble are less soluble than this.

15 Some highly water-soluble materials which may be used are listed below, with their solubilities expressed as grams of solid to form a saturated solution in 100 grams of water at 20°C.

<u>Material</u>	<u>Water Solubility (g/100g)</u>
Sodium citrate dihydrate	72
20 Potassium carbonate	112
Urea	>100
Sodium acetate	119
Sodium acetate trihydrate	76
Magnesium sulphate 7H <sub>2</sub> O	71

By contrast the solubilities of some other common materials at 20°C are:

<u>Material</u>	<u>Water Solubility (g/100g)</u>
Sodium chloride	36
5      Sodium sulphate decahydrate	21.5
Sodium carbonate anhydrous	8.0
Sodium percarbonate anhydrous	12
Sodium perborate anhydrous	3.7
Sodium tripolyphosphate anhydrous	15

10      Preferably this highly water soluble material is incorporated as particles of the material in a substantially pure form (i.e. each such particle contains over 95% by weight of the material). However, the said particles may contain material of such solubility in a 15 mixture with other material, provided that material of the specified solubility provides at least 40% by weight of these particles.

20      It may be preferred that the highly water-soluble material is a salt which dissolves in water in an ionised form. As such a salt dissolves it leads to a transient local increase in ionic strength which can assist disintegration of the tablet by preventing nonionic surfactant from swelling and inhibiting dissolution of other materials.

Another possibility is that the water-swellable polymer is

accompanied by further particles to promote disintegration which contain sodium tripolyphosphate with at least 40% (by weight of the further particles) of the anhydrous phase I form.

5      Sodium tripolyphosphate is very well known as a sequestering builder in detergent compositions. It exists in a hydrated form and two crystalline anhydrous forms. These are the normal crystalline anhydrous form, known as phase II which is the low temperature form, and phase I which is stable at high temperature. The conversion of phase II to phase I proceeds fairly rapidly on heating above the transition temperature, which is about 420°C, but the reverse reaction is slow. Consequently phase I sodium tripolyphosphate is metastable at ambient temperature.

15     A process for the manufacture of particles containing a high proportion of the phase I form of sodium tripolyphosphate by spray drying below 420°C is given in US-A-4536377.

20     Particles which contain this phase I form will often contain the phase I form of sodium tripolyphosphate as at least 50% or 55% by weight of the tripolyphosphate in the particles.

Suitable material is commercially available. Suppliers include Rhone-Poulenc, France and Albright & Wilson, UK.

Another possibility is that further particles which promote disintegration contain at least 40% (of their own weight) of sodium tripolyphosphate which is partially hydrated.

5 The extent of hydration may lie in a range from 0.5 to 4% by weight of the sodium tripolyphosphate in the particles, or it may be higher. Indeed fully hydrated sodium tripolyphosphate may be used to provide these particles.

10 It is possible that the particles contain at least 40 wt% sodium tripolyphosphate which has a high phase I content but is also sufficiently hydrated so as to contain at least 0.5% water by weight of the sodium tripolyphosphate.

Particles as above may provide sodium tripolyphosphate in a quantity which is at least 8%, e.g. 8 to 30%, by weight of the composition of the first said region(s) of the tablet.

15 The remainder of the tablet composition used to form the first said region(s) of the tablet may include additional sodium tripolyphosphate. This may be in any form, including sodium tripolyphosphate with a high content of the anhydrous phase II form.

20 It is within the scope of the second aspect of this invention that the first said region includes particles containing a highly water soluble compound or particles containing phase I tripolyphosphate or hydrated tripolyphosphate, all as discussed above, without any

water-swellable polymer.

Surfactant Compounds

Compositions which are compacted to form some tablet regions will contain one or more detergent surfactants. In

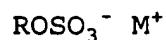
5 a fabric washing composition, these preferably provide from 5 to 50% by weight of the overall tablet composition, more preferably from 8 or 9% by weight of the overall composition up to 40% or 50% by weight. Surfactant may be anionic (soap or non-soap), cationic, zwitterionic, 10 amphoteric, nonionic or a combination of these.

Anionic surfactant may be present in an amount from 0.5 to 50% by weight, preferably from 2% or 4% up to 30% or 40% by weight of the tablet composition.

Synthetic (i.e. non-soap) anionic surfactants are well

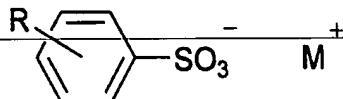
15 known to those skilled in the art. Examples include alkylbenzene sulphonates, particularly sodium linear alkylbenzene sulphonates having an alkyl chain length of C<sub>8</sub>-C<sub>15</sub>; olefin sulphonates; alkane sulphonates; dialkyl sulphosuccinates; and fatty acid ester sulphonates.

20 Primary alkyl sulphate having the formula



in which R is an alkyl or alkenyl chain of 8 to 18 carbon

atoms especially 10 to 14 carbon atoms and  $M^+$  is a solubilising cation, is commercially significant as an anionic surfactant. Linear alkyl benzene sulphonate of the formula



5 where R is linear alkyl of 8 to 15 carbon atoms and  $M^+$  is a solubilising cation, especially sodium, is also a commercially significant anionic surfactant.

Frequently, such linear alkyl benzene sulphonate or primary alkyl sulphate of the formula above, or a mixture thereof 10 will be the desired anionic surfactant and may provide 75 to 100 wt% of any anionic non-soap surfactant in the composition.

In some forms of this invention the amount of non-soap 15 anionic surfactant lies in a range from 5 to 20 wt% of the tablet composition.

It may also be desirable to include one or more soaps of fatty acids. These are preferably sodium soaps derived from naturally occurring fatty acids, for example, the fatty acids from coconut oil, beef tallow, sunflower or 20 hardened rapeseed oil.

Suitable nonionic surfactant compounds which may be used

include in particular the reaction products of compounds having a hydrophobic group and a reactive hydrogen atom, for example, aliphatic alcohols, acids, amides or alkyl phenols with alkylene oxides, especially ethylene oxide.

5 Specific nonionic surfactant compounds are alkyl (C<sub>8-22</sub>) phenol-ethylene oxide condensates, the condensation products of linear or branched aliphatic C<sub>8-20</sub> primary or secondary alcohols with ethylene oxide, and products made by condensation of ethylene oxide with the reaction

10 products of propylene oxide and ethylene-diamine.

Especially preferred are the primary and secondary alcohol ethoxylates, especially the C<sub>9-11</sub> and C<sub>12-15</sub> primary and secondary alcohols ethoxylated with an average of from 5 to 20 moles of ethylene oxide per mole of alcohol.

15 In some fabric washing tablets of this invention, the amount of nonionic surfactant lies in a range from 4 to 40%, better 4 or 5 to 30% by weight of the whole tablet.

Many nonionic surfactants are liquids. These may be absorbed onto particles of the composition.

20 In a machine dishwashing tablet the surfactant may be wholly nonionic, in an amount below 5 wt% of the whole tablet although it is known to include some anionic surfactant and to use up to 10 wt% surfactant in total.

Detergency Builder

A composition which is compacted to form some regions of tablets will contain from 5 to 80%, more usually 15 to 60%

by weight of detergency builder. This may be provided

5 wholly by water soluble materials, or may be provided in

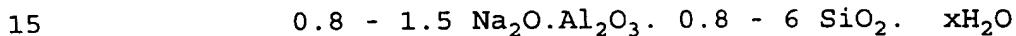
large part or even entirely by water-insoluble material

with water-softening properties. Water-insoluble

detergency builder may be present as 5 to 80 wt%, better 5

to 60 wt% of the composition.

10 Alkali metal aluminosilicates are strongly favoured as environmentally acceptable water-insoluble builders for fabric washing. Alkali metal (preferably sodium) aluminosilicates may be either crystalline or amorphous or mixtures thereof, having the general formula:



These materials contain some bound water (indicated as "xH<sub>2</sub>O") and are required to have a calcium ion exchange capacity of at least 50 mg CaO/g. The preferred sodium aluminosilicates contain 1.5-3.5 SiO<sub>2</sub> units (in the formula above). Both the amorphous and the crystalline materials can be prepared readily by reaction between sodium silicate and sodium aluminate, as amply described in the literature.

Suitable crystalline sodium aluminosilicate ion-exchange detergency builders are described, for example, in GB

25 1429143 (Procter & Gamble). The preferred sodium

aluminosilicates of this type are the well known commercially available zeolites A and X, the novel zeolite P described and claimed in EP 384070 (Unilever) and mixtures thereof.

5 Conceivably a water-insoluble detergency builder could be a layered sodium silicate as described in US 4664839. NaSKS-6 is the trademark for a crystalline layered silicate marketed by Hoechst (commonly abbreviated as "SKS-6"). 10 NaSKS-6 has the delta- $\text{Na}_2\text{SiO}_5$  morphology form of layered silicate. It can be prepared by methods such as described in DE-A-3,417,649 and DE-A-3,742,043. Other such layered silicates, such as those having the general formula 15  $\text{NaMSi}_x\text{O}_{2x+1} \cdot y\text{H}_2\text{O}$  wherein M is sodium or hydrogen, x is a number from 1.9 to 4, preferably 2, and y is a number from 0 to 20, preferably 0 can be used.

Water-soluble phosphorous-containing inorganic detergency builders, include the alkali-metal orthophosphates, metaphosphates, pyrophosphates and polyphosphates. Specific examples of inorganic phosphate builders include 20 sodium and potassium tripolyphosphates, orthophosphates and hexametaphosphates.

Non-phosphorous water-soluble builders may be organic or inorganic. Inorganic builders that may be present include alkali metal (generally sodium) carbonate; while organic 25 builders include polycarboxylate polymers, such as

polyacrylates, acrylic/maleic copolymers, and acrylic phosphonates, monomeric polycarboxylates such as citrates, gluconates, oxydisuccinates, glycerol mono- di- and trisuccinates, carboxymethyloxysuccinates, 5 carboxymethyloxymalonates, dipicolinates and hydroxyethyliminodiacetates.

At least one region of a fabric washing tablet preferably include polycarboxylate polymers, more especially polyacrylates and acrylic/maleic copolymers which can 10 function as builders and also inhibit unwanted deposition onto fabric from the wash liquor.

#### Bleach System

Tablets according to the invention may contain a bleach system in at least one region of a tablet. This preferably 15 comprises one or more peroxy bleach compounds, for example, inorganic persalts or organic peroxyacids, which may be employed in conjunction with activators to improve bleaching action at low wash temperatures. If any peroxygen compound is present, the amount is likely to lie 20 in a range from 10 to 25% by weight of the composition.

Preferred inorganic persalts are sodium perborate monohydrate and tetrahydrate, and sodium percarbonate, advantageously employed together with an activator. Bleach activators, also referred to as bleach precursors, have 25 been widely disclosed in the art. Preferred examples

include peracetic acid precursors, for example,

tetraacetylene diamine (TAED), now in widespread

commercial use in conjunction with sodium perborate; and

perbenzoic acid precursors. The quaternary ammonium and

5 phosphonium bleach activators disclosed in US 4751015 and

US 4818426 (Lever Brothers Company) are also of interest.

Another type of bleach activator which may be used, but

which is not a bleach precursor, is a transition metal

catalyst as disclosed in EP-A-458397, EP-A-458398 and EP-A-

10 549272. A bleach system may also include a bleach

stabiliser (heavy metal sequestrant) such as

ethylenediamine tetramethylene phosphonate and

diethylenetriamine pentamethylene phosphonate.

As indicated above, if a bleach is present and is a water-

15 soluble inorganic peroxygen bleach, the amount may well be

from 10% to 25% by weight of the composition.

#### Other Detergent Ingredients

The detergent tablets of the invention may also contain one

of the detergency enzymes well known in the art for their

20 ability to degrade and aid in the removal of various soils

and stains. Suitable enzymes include the various

proteases, cellulases, lipases, amylases, and mixtures

thereof, which are designed to remove a variety of soils

and stains from fabrics. Examples of suitable proteases

25 are Maxatase (Trade Mark), as supplied by Gist-Brocades

N.V., Delft, Holland, and Alcalase (Trade Mark), and

Savinase (Trade Mark), as supplied by Novo Industri A/S, Copenhagen, Denmark. Detergency enzymes are commonly employed in the form of granules or marumes, optionally with a protective coating, in amount of from about 0.1% to 5 about 3.0% by weight of the composition; and these granules or marumes present no problems with respect to compaction to form a tablet.

The detergent tablets of the invention may also contain a fluorescer (optical brightener), for example, Tinopal (Trade Mark) DMS or Tinopal CBS available from Ciba-Geigy AG, Basel, Switzerland. Tinopal DMS is disodium 4,4'-bis-(2-morpholino-4-anilino-s-triazin-6-ylamino) stilbene disulphonate; and Tinopal CBS is disodium 2,2'-bis-(phenyl-styryl) disulphonate.

15 An antifoam material is advantageously included, especially if a detergent tablet is primarily intended for use in front-loading drum-type automatic washing machines. Suitable antifoam materials are usually in granular form, such as those described in EP 266863A (Unilever). Such 20 antifoam granules typically comprise a mixture of silicone oil, petroleum jelly, hydrophobic silica and alkyl phosphate as antifoam active material, absorbed onto a porous absorbed water-soluble carbonate-based inorganic carrier material. Antifoam granules may be present in an 25 amount up to 5% by weight of the composition.

It may also be desirable that a detergent tablet of the invention includes an amount of an alkali metal silicate, particularly sodium ortho-, meta- or disilicate. The presence of such alkali metal silicates at levels, for 5 example, of 0.1 to 10 wt%, may be advantageous in providing protection against the corrosion of metal parts in washing machines, besides providing some measure of building and giving processing benefits in manufacture of the particulate material which is compacted into tablets.

10 A tablet for fabric washing will generally not contain more than 15 wt% silicate. A tablet for machine dishwashing will often contain more than 20 wt% silicate.

15 Further ingredients which can optionally be employed in a region of a fabric washing detergent tablet of the invention include anti-redeposition agents such as sodium carboxymethylcellulose, straight-chain polyvinyl pyrrolidone and the cellulose ethers such as methyl cellulose and ethyl hydroxyethyl cellulose, fabric-softening agents; heavy metal sequestrants such as EDTA; 20 perfumes; and colorants or coloured speckles.

Particle Size and Distribution

Each discrete region of a detergent tablet of this invention, is a matrix of compacted particles.

Preferably the particulate composition has an average

particle size in the range from 200 to 2000  $\mu\text{m}$ , more

preferably from 250 to 1400  $\mu\text{m}$ . Fine particles, smaller than 180  $\mu\text{m}$  or 200  $\mu\text{m}$  may be eliminated by sieving before tabletting, if desired, although we have observed that this

5 is not always essential.

While the starting particulate composition may in principle have any bulk density, the present invention is especially relevant to tablets made by compacting powders of relatively high bulk density, because of their greater

10 tendency to exhibit disintegration and dispersion problems.

Such tablets have the advantage that, as compared with a tablet derived from a low bulk density powder, a given dose of composition can be presented as a smaller tablet.

Thus the starting particulate composition may suitably have

15 a bulk density of at least 400 g/litre, preferably at least 500 g/litre, and perhaps at least 600 g/litre.

Granular detergent compositions of high bulk density prepared by granulation and densification in a high-speed mixer/granulator, as described and claimed in EP 340013A  
20 (Unilever), EP 352135A (Unilever), and EP 425277A (Unilever), or by the continuous granulation/densification processes described and claimed in EP 367339A (Unilever) and EP 390251A (Unilever), may be used to product at least some tablet regions.

When a tablet incorporates particles of a water-swellable polymeric material, or other particles to promote disintegration, they are preferably mixed with the remainder of the particulate composition, prior to 5 compaction into tablets.

Manufacture of a tablet with two layers of differing composition may be carried out by placing a predetermined quantity of one composition in a mould, then adding a second composition on top, and next driving a die into the 10 mould to cause compaction.

Alternatively, a predetermined quantity of a composition may be placed in a mould and compacted by driving a die into the mould, followed by removing the die, adding a second composition and compacting again.

15 Tablets with even more layers can be made by these routes, but with extra stages of loading particulate material into the die, and possibly compacting after each stage.

Tableting machinery able to carry out such operations is known, for example suitable tablet presses are available 20 from Fette and from Korch.

Tableting may be carried out at ambient temperature or at a temperature above ambient which may allow adequate strength to be achieved with less applied pressure during

compaction. In order to carry out the tableting at a temperature which is above ambient, the particulate composition is preferably supplied to the tableting machinery at an elevated temperature. This will of course supply heat to the tableting machinery, but the machinery may be heated in some other way also.

If any heat is supplied, it is envisaged that this will be supplied conventionally, such as by passing the particulate composition through an oven, rather than by any application of microwave energy.

The size of a tablet will suitably range from 10 to 160 grams, preferably from 15 to 60 g, depending on the conditions of intended use, and whether it represents a dose for an average load in a fabric washing or dishwashing machine or a fractional part of such a dose. The tablets may be of any shape. However, for ease of packaging they are preferably blocks of substantially uniform cross-section, such as cylinders or cuboids. The overall density of a tablet preferably lies in a range from 1040 or 15 1050gm/litre up to 1300gm/litre. The tablet density may 20 well lie in a range up to no more than 1250 or even 1200gm/litre.

Example 1demonstrating the effect of water-swellable polymer

Experiments were carried out with a polymeric material derived from cellulose and marketed by Rettenmaier GmbH as 5 "Arbocel A1". As supplied it has particles with a range of shapes and particle sizes (as determined by sieve analysis) with an average diameter of 1mm. It was found to have a water-uptake of 5.7 gm/gm.

10 The material was mixed, at a concentration of 5% by weight with each of four detergent powders. These powders were then stamped into detergent tablets. Control tablets were made from the same powders without Arbocel A1. The main constituents of these powders are given in the table below.

15 Some tablets made from each of the four powders were fully immersed in water at 20°C. The tablets containing Arbocel were observed to break up in times less than one minute. During the same period of time the control tablets remain intact.

20 For some of the tablets the break-up, dispersion and dissolution of tablets was measured by a test procedure in which a tablet is placed on a plastic sieve with 2mm mesh size which was immersed in 9 litres of demineralised water at ambient temperature of 20°C. The water conductivity was monitored until it reached a constant value. The time for 25 dissolution of the tablets was taken as the time ( $T_{90}$ ) for

change in the water conductivity to reach 90% of its final magnitude. The results are included in the table below.

5	Composition of Powder	Powder bulk density	Visible disintegration		T <sub>90</sub> conductivity measurement	
			without Arbocel A1	with Arbocel A1	without Arbocel A1	with Arbocel A1
	A 16 wt% total surfactant, 46% sodium tripolyphosphate	640 gm/litre	>10 minutes	<1 minute	4 minutes	2 minutes
	B 16 wt% total surfactant, 31% zeolite, zero phosphate	880 gm/litre	>10 minutes	<1 minute	over 10 minutes	2 minutes
	C 19 wt% total surfactant, 15% zeolite, 10% layered silicate, zero phosphate		>10 minutes	<1 minute	over 10 minutes	4 minutes
	D spray dried: 9% total surfactant, 35% sodium tripolyphosphate	about 550 gm/litre	>10 minutes	<1 minute		

10 In comparative experiments, tablets were made using 5% of Arbocel A1 which had been gently ground with a pestle and mortar to reduce the size of the particles, (to the primary particle size of approximately 120 micrometres). This ground material was much less effective at promoting tablet disintegration.

Example 2 demonstrating particles to provide disintegration

Tablets for use in fabric washing were made, starting with a spray-dried base powder of the following composition:

	Ingredient	Parts by Weight
5	Sodium linear alkylbenzene sulphonate	11.0
	Sodium tripolyphosphate (Added to the slurry as anhydrous sodium tripolyphosphate containing at least 70% phase II form.)	16.8
10	C <sub>13-15</sub> fatty alcohol 7EO	2.4
	C <sub>13-15</sub> fatty alcohol 3EO	2.3
	Sodium silicate	4.0
	Soap	0.21
	Acrylate/maleate copolymer	1.5
15	Sodium sulphate, moisture and minor ingredients	balance to 45 parts

A number of particulate compositions were made by mixing this powder with other ingredients as tabulated below these included particles of sodium tripolyphosphate specified to contain 70% phase I form and contain 3.5% water of hydration (Rhodia-Phos HPA 3.5 available from Rhone-Poulenc).

The added ingredients also included particles of water-insoluble water-swellable polymeric material. This material was "Arbocel A1" as in Example 1. For some compositions this material was sieved to provide a fraction with a narrower range of particle size.

The compositions were balanced to 100% by including varying

amounts of dense anhydrous sodium carbonate.

The various compositions contained the following percentages by weight:

	Ingredient	% by weight
5	Base powder	45.0
	Sodium percarbonate granules	15.0
	TAED granules	3.4
	Anti-foam granules	3.2
10	Perfume, enzymes and other minor ingredients	3.5
	HPA tripolyphosphate	variable, 15 to 30%
	Water-swellable polymer	variable, 0 to 5%
	Sodium carbonate	balance, 0 to 12%

40g portions of each composition were made into cylindrical tablets of 44 mm diameter, using a Fette pilot plant press, with a fixed level of applied pressure so as to produce tablets of density in a range from 1100 to 1250kg/m<sup>3</sup>

The strength of these tablets was measured using an Instron universal testing machine to compress a tablet until fracture. The value of diametral fracture stress (DFS) was then calculated using the equation

$$\sigma = \frac{2P}{\pi Dt}$$

where  $\sigma$  is the diametral fracture stress in Pascals, P is the applied load in Newtons to cause fracture, D is the tablet diameter in metres and t is the tablet thickness in metres.

The break-up, dispersion and dissolution of tablets was measured by a test procedure as in Example 1.

The percentages of HPA tripolyphosphate, and polymeric material, together with the DFS values and conductivity

5 results are set out in the following table:

code #	HPA tripoly-phosphate	polymeric material	balancing carbonate	DFS (kPa)	T <sub>90</sub> (minute)
A	30%	0	0	43	3.0
B	24%	2% as supplied	4%	32	3.2
C	15%	5% as supplied	10%	18	<1
D	15%	0	15%	45	>9.0
E	15%	5% 470-800 $\mu$	10.0	30	3.2
F	15%	5% 800-1400 $\mu$	10.0	21	1.4
G	15%	3% 800-1400 $\mu$	12.0	33	2.8

Example 3

An enzyme-containing composition was prepared from the base powder used in Example 2, and added ingredients as follows:

5	Ingredient	% by weight
	Base powder	45.0
	Protease (Savinase ex Novo)	4.0
	Lipase (Lipolase ex Novo)	0.8
	Amylase (Termamyl ex Novo)	2.5
10	Anti-foam granules	3.2
	Minor ingredients	2.0
	HPA tripolyphosphate	22.5
	Water-swellable polymer	5
	Sodium carbonate	15
	TOTAL	100

15 15gm of this composition was used to provide one layer of a two layer tablet. The other layer of the tablet was provided by 35gm of a bleach containing composition as follows:

20	Ingredient	% by weight
	Base powder	45.0
	Sodium percarbonate granules	20.0
	TAED granules	5.0
	Anti-foam granules	3.0
	Minor ingredients	2.0
25	HPA tripolyphosphate	15
	Sodium carbonate	10
	TOTAL	100

On addition to water the first layer disintegrated and its

soluble constituents dissolved in under one minute. The second layer remained intact for about 9 minutes.

Example 4

Two granulated base powders have the following

5 compositions:

Ingredient	parts by weight	
	A	B
Sodium linear alkylbenzene sulphonate	4.0	10
C <sub>13-15</sub> fatty alcohol 7EO.	6.7	1.7
C <sub>13-15</sub> fatty alcohol 3EO.	4.1	2.4
10 Soap	0	1.0
Zeolite A24	28.0	21.0
Sodium carbonate	3.7	3.7
Sodium citrate dihydrate	4.6	3.1
moisture and minors	5.4	5.1
15 TOTAL	56.5	48

Two particulate compositions were made by mixing these powders with other ingredients as set out in the table below. The water-swellable polymer was "Arbocel A1" used as supplied.

5	Ingredient	% by weight	
		C	D
	Base powder A	56.5	
	Base powder B		48
	Sodium perborate monohydrate	8	14.8
	TAED granules	2.3	5.5
10	Anti-foam granules	2.0	2.0
	Fluorescer granules	0	2.0
	Sodium silicate granules	3.7	3.7
	Acrylate/maleate copolymer	3.0	1.0
15	Perfume, enzymes and other minor ingredients	3.5	3.5
	Sodium acetate trihydrate	16	11.0
	Water-swellable polymer	5	2.5
	Sodium carbonate	0	6.0
	TOTAL	100	100

20 Tablets are made using 20g of composition C to make one layer and 20g of composition D to make the other layer. On addition to water the layer of composition A disintegrates within less one minute, releasing builder to substantially soften the water before the disintegration of the second 25 layer after about six minutes, which releases most of the anionic surfactant, including all of the soap.

Example 5

Two particulate compositions are made by mixing base powder B from the previous example with other ingredients as set out in the table below. The water-swellable polymer was 5 "Arbocel A1" used as supplied. The soil suspending polymer was a graft copolymer as described in US-A-4746456.

	Ingredient		% by weight	
		E	F	
10	Base powder B	48	48	
	Sodium perborate monohydrate	13.9	13.9	
	TAED granules	5.3	5.3	
	Anti-foam granules	2.0	2.0	
	Fluorescer granules	2.0	2.0	
	Sodium silicate granules	3.7	3.7	
	Acrylate/maleate copolymer	1.5	1.0	
	Polyvinylpyrrolidone	0	1.5	
	Soil release polymer	0	4.5	
	Perfume	0.6	0.6	
15	enzymes	2.0	0	
	Sodium acetate trihydrate	16	11.0	
	Water-swellable polymer	5	2.0	
	Sodium carbonate	0	4.5	
	<b>TOTAL</b>	<b>100</b>	<b>100</b>	

Tablets are made using 30g of composition E to make one layer and 15g of composition F to make the other layer. On 25 addition to water the layer of composition E disintegrates within one minute. The layer of composition F disintegrates after about 7 minutes, thus delaying release of the two polymers.

CLAIMS

1. A tablet of compacted particulate cleaning composition containing surfactant and detergency builder, and which has a plurality of discrete regions with
- 5 differing compositions, characterised in that at least one said region of the tablet contains water-swellable polymer in a concentration greater than in at least one other region of the tablet, to promote disintegration of the first said region(s) before said other(s).
- 10 2. A tablet according to claim 1 wherein the first said region(s) contain enzyme(s) and the said other region(s) contain bleach or bleach activator.
- 15 3. A tablet according to claim 1 or claim 2, wherein the polymeric material has a particle dimension of at least 400 micrometres.
4. A tablet according to claim 1 or claim 2, wherein the polymeric material has a particle dimension of at least 500 micrometres.
- 20 5. A tablet according to any one of claims 1 to 4, wherein the polymeric material is substantially nonionic such that the charge density of the polymeric material does not exceed  $10^{-3}$ .

6. A tablet according to any one of the preceding claims, wherein the polymeric material is a polysaccharide.

7. A tablet according to any one of the preceding claims, wherein the first said region(s) also contain

5 further particles comprising at least 40% (by weight of the particles) of one or more materials selected from the group consisting of

- compounds with a water-solubility exceeding 50 grams per 100 grams water
- 10 • phase 1 sodium tripolyphosphate
- sodium tripolyphosphate which is partially hydrated so as to contain water of hydration in an amount which is at least 0.5% by weight of the sodium tripolyphosphate in the particles.

15 8. A tablet according to any one of the preceding claims, wherein the first said region(s) of the tablet contain 0.5 to 50 wt% surfactant, 5 to 80 wt% detergency builder and 0.1 to 8 wt% of said particles containing water-insoluble, water-swellable polymeric material.

20 9. A tablet according to claim 8, wherein the first said region(s) of the tablet contain from 5 to 25 wt% of further particles as specified in claim 7 to promote disintegration.

10. A tablet of compacted particulate detergent

composition containing surfactant and detergency builder,  
which has a plurality of discrete regions with differing  
compositions, characterised that at least one region of the  
tablet has a composition which causes it to disintegrate  
5 ahead of at least one other region of the tablet,

wherein the first said region contains enzyme in  
a quantity greater than in the other said region, and the  
other said region contains surfactant and/or bleach in a  
greater quantity than in the first said region.

10 11. A tablet according to any one of the preceding  
claims, which overall contains from 5 to 50% by weight of  
surfactant and 5 to 80% by weight of detergency builder.

12. A tablet according to claim 11, which overall  
contains 5 to 60 wt% water-insoluble detergency builder.

15 13. A tablet according to claim 11, which overall  
contains from 10 to 80% by weight of water-soluble  
detergency builder.

14. A process of making a tablet according to any one  
of claims 1 to 12, comprising placing quantities of at  
20 least two particulate compositions in a mould and  
compacting these therein, characterised by adding water-  
swellable polymeric material to one of the compositions  
prior to compaction, so as to establish therein a greater  
concentration of the water-swellable polymeric material

than in the other composition(s).

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15. Process according to claim 14 wherein the swellable polymeric material is added to the other particulate ingredients as particles which contain at least 75% of  
5 their own weight of the polymeric material.

# INTERNATIONAL SEARCH REPORT

International Application No  
PCT/EP 98/03491

## A. CLASSIFICATION OF SUBJECT MATTER

C 11 D 17/06, C 11 D 3/37, C 11 D 17/00

According to International Patent Classification (IPC) or to both national classification and IPC6

## B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

C 11 D

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

## C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	EP 0711828 A2 (UNILEVER PLC) 15 May 1996 (15.05.96), the whole document. --	1-15
A	EP 0466484 A2 (UNILEVER PLC) 15 January 1992 (15.01.92), examples. --	1-15
A	EP 0716144 A2 (UNILEVER PLC) 12 June 1996 (12.06.96), claims. --	1-15
A	EP 0481793 A1 (UNILEVER PLC) 22 April 1992 (22.04.92), the whole document	1-15

Further documents are listed in the continuation of box C.

Patent family members are listed in annex.

### \* Special categories of cited documents :

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Date of the actual completion of the international search

06 October 1998

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## INTERNATIONAL SEARCH REPORT

International Application No

PCT/EP 98/03491

-2-

**C(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT**

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	<p style="text-align: center;">(cited in the application).</p> <p style="text-align: center;">--</p> <p>EP 0522766 A2            (UNILEVER PLC) 13 January            1993 (13.01.93),            claims.</p> <p style="text-align: center;">-----</p>	1-15

**ANHANG****ANNEX****ANNEXE**

zum internationalen Recherchenbericht über die internationale Patentanmeldung Nr.

to the International Search Report to the International Patent Application No.

au rapport de recherche international relatif à la demande de brevet international n°

PCT/EP 98/03491 SAE 202737

In diesem Anhang sind die Mitglieder der Patentfamilien der im obengenannten internationalen Recherchenbericht angeführten Patentdokumente angegeben. Diese Angaben dienen nur zur Orientierung und erfolgen ohne Gewähr.

This Annex lists the patent family members relating to the patent documents cited in the above-mentioned international search report. The Office is in no way liable for these particulars which are given merely for the purpose of information.

La présente annexe indique les membres de la famille de brevets relatifs aux documents de brevets cités dans le rapport de recherche international visée ci-dessus. Les renseignements fournis sont donnés à titre indicatif et n'engagent pas la responsabilité de l'Office.

Im Recherchenbericht angeführtes Patentdokument Patent document cited in search report Document de brevet cité dans le rapport de recherche	Datum der Veröffentlichung Publication date Date de publication	Mitglied(er) der Patentfamilie Patent family member(s) Membre(s) de la famille de brevets	Datum der Veröffentlichung Publication date Date de publication
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